

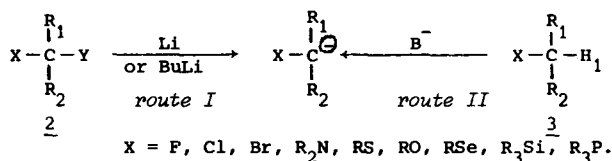
REACTIVITY OF MIXED SULFO-SELENO ACETALS AND OXO-SELENO ACETALS
WITH N-BUTYLLITHIUM : A NEW ROUTE TO SUBSTITUTED EPOXIDES (1).

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α -Heterosubstituted carbanions 1 are highly reactive intermediates in organic synthesis as carbenoids or carbenes. These reagent 1 are well known if $R_1 = R_2 = H$ and are generally prepared via halogen-metal exchange from α -hetero substituted alkyl halides 2 : route I or by action of a suitable base on compounds 3 : route II (Scheme I).

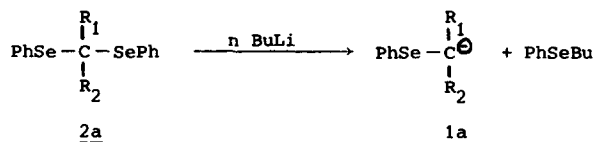
Scheme I



Nevertheless higher homologs of 1 ($R_1 = H$, Alkyl ; $R_2 = Alkyl$) are much less well known (2) due to the low acidity of H_1 in 3 which makes route II difficult and attempted application of route I leads to an olefin by β elimination reaction or to other side reactions.

We (3) and others (4) recently achieved the first synthesis of highly substituted α -seleno carbanions 1a via the easy cleavage of the C-Se bond in selenoacetals 2a with n-butyllithium (Scheme II).

Scheme II



The present communication reports the behaviour of mixed, sulfo-selenoacetals 2b and oxo-selenoacetals 2c with n-butyllithium. The generation of unknown highly substituted α -sulfocarbanion 1b (1-4) is described, as well as their use for a convenient synthesis of epoxide (5,6).

For instance the sulfo-selenoacetals 2b (7) are readily cleaved by n-butyllithium in THF (-78° C, 1 hr) to give butyl phenyl selenide and α -sulfo-carbanions 1b. The highly nucleophilic α -sulfo-carbanion 1b adds a variety of carbonyl compounds, even highly enolizable ones such as deoxybenzoine, producing the β -hydroxysulfides 5 in fair yield (scheme III, table I).

Scheme III

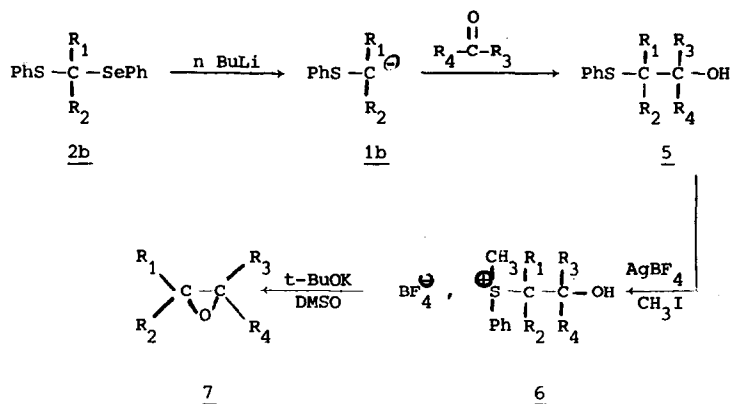


Table I

R ₁	R ₂	R ₃	R ₄	β -hydroxysulfide <u>5</u>	Epoxide <u>7</u>
H	H	H	C ₆ H ₅	94 %	55 %
H	H	H	nC ₆ H ₁₃	84	50
H	H	-(CH ₂) ₂ -CH(tC ₄ H ₉)-(CH ₂) ₂ -		58	60
H	CH ₃	C ₆ H ₅	CH ₂ -C ₆ H ₅	92	60
H	CH ₃	-CH ₂ - (CH ₂) ₃ - CH ₂ -		56	60
H	CH ₃	H	C ₆ H ₅	86	45
H	CH ₃	H	nC ₆ H ₁₃	45	34
CH ₃	CH ₃	H	nC ₆ H ₁₃	57	
CH ₃	CH ₃	C ₆ H ₅	CH ₂ -C ₆ H ₅	30	
CH ₃	CH ₃	-CH ₂ - (CH ₂) ₃ - CH ₂ -		51	

yields of 5 and 6 refer to purified products and spectroscopic data are in accord with the proposed structures.

It should be mentioned that the parent α -sulfo-carbanion ($R_1-R_2=H$) was previously prepared from methyl phenyl sulfide by hydrogen abstraction with *n*-butyllithium (6,8). However the same procedure cannot be applied to higher homologue ($R_1 = \text{alkyl}$) due to metalation of the phenyl sulfide in the ortho-position (9).

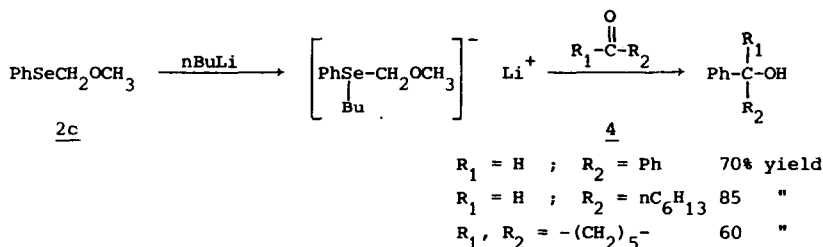
As synthetic application of the C-Se bond cleavage in mixed sulfo selenoacetals we describe the connective (7) synthesis (Scheme III, table I) of epoxides, including some of them which cannot be produced by conventional connective methods (6,10) such as 1-benzyl-1-phenyl-2-methyloxirane.

β -Hydroxysulfonium salts 6 are readily formed when β -hydroxysulfides 5 are treated with methyl iodide in the presence of silver tetrafluoroborate (neat 25° C, 2 hrs); subsequent treatment of the crude salt with base (tBuOK-DMSO, 25° C, 2 hrs) produces the desired epoxide 7 and thioanisole.

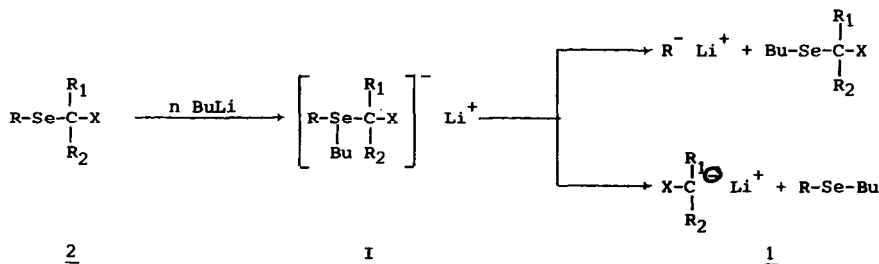
Unfortunately, rearranged carbonyl compounds and unidentified products (11) are obtained when β -hydroxysulfides ($R_1, R_2 = \text{alkyl}$) are treated under our conditions ($\text{CH}_3\text{I-AgBF}_4$) and tetra-substituted epoxides are not yet accessible by this way.

Finally, we studied the reaction of *n*-butyllithium with the oxo-selene acetal 2c (THF, -78° C, 1 hr). In this case, we observe the preponderant cleavage of the Se-C bond leading to phenyllithium rather than to the α -alkoxy carbanion 1c as shown by the phenylated products obtained by adding carbonyl compounds in the medium (Scheme IV).

Scheme IV



We are not yet able to determine the factors which decide the mode of cleavage of the selenides 2 by *n*-butyllithium; following an hypothesis the complex I decomposes to give the best stabilised carbanion. The *d* orbital stabilisation should be of great importance.



Work is actually in progress to determine these modes of cleavage in compound 2 ($Y = RSe$; $X = SiR_3, NR_2, PR_2$ and halogens). Nevertheless the C-Se bond cleavage should provide an original and potent way of carbanion generation.

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